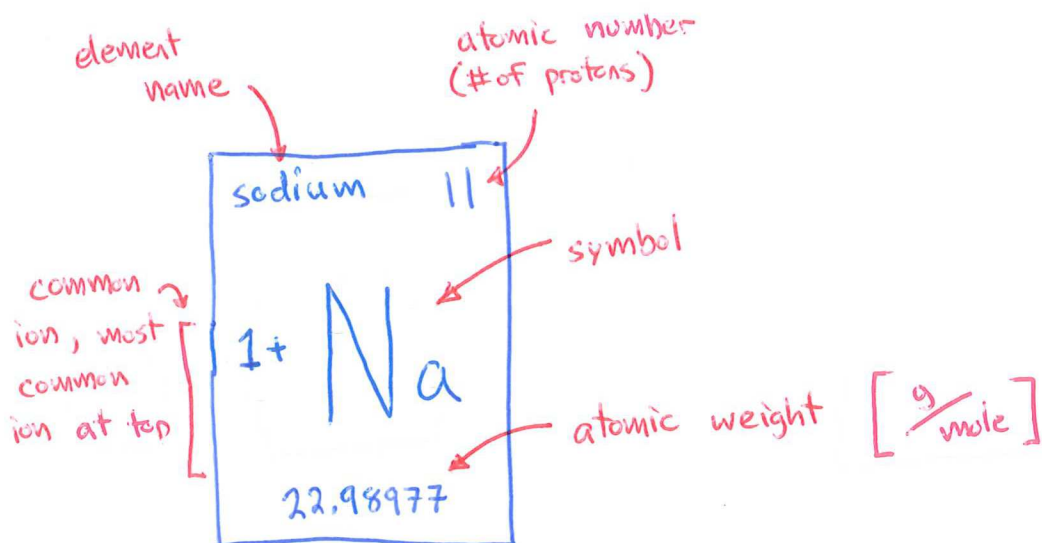


# Formula Weight (FW)



• atomic weight is for a single element [ $\frac{g}{mole}$ ]  
if multiple ( $N_2, O_2, \dots$ ) multiply by subscript.

• Formula weight (FW) or molecular weight is for a compound ( $NaCl, H_2O, MgSO_4, \dots$ ) add [ $\frac{g}{mole}$ ]  
all atomic weights.

$$FW = \sum \text{atomic weights}_i$$

atomic weights of compound parts

examples:

$$FW_{H_2O} = (2)(1.0079) + (15.9994) = 18.0152 \frac{g}{mole}$$

molecular weight of H

molecular weight of O

$$FW_{NaCl} = (22.9898) + (35.453) = 58.4428 \frac{g}{mole}$$

$$FW_{MgSO_4} = (24.305) + (32.065) + (4)(15.9994) = 120.3676 \frac{g}{mole}$$